

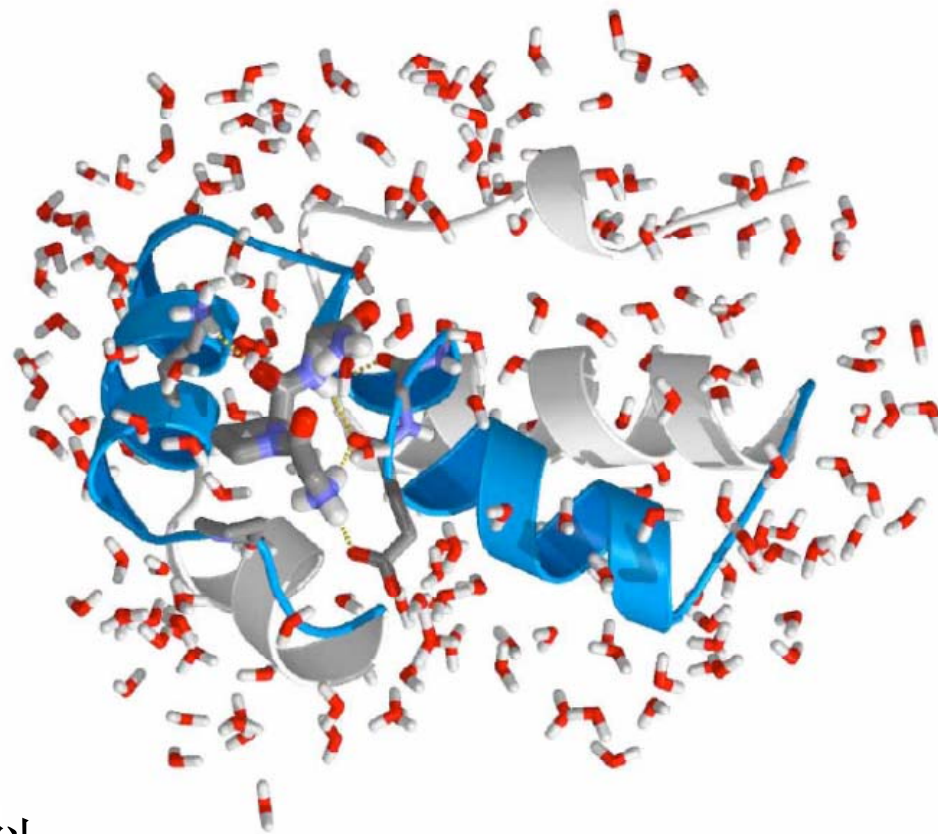
# 前沿化学实验13——理论与计算化学

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分子动态与稳态结构国家重点实验室

## 3. 分子动力学模拟简介

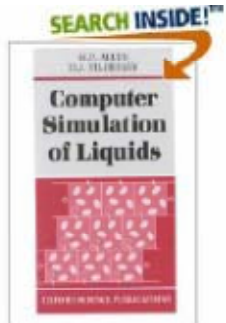


**Gromacs  
Logo**

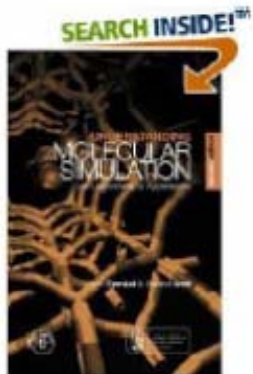
主要参考资料:

**Emppu Salonen Helsinki University of Technology  
Molecular Dynamics Simulation - Michel Cuendet - EMBL 2008**

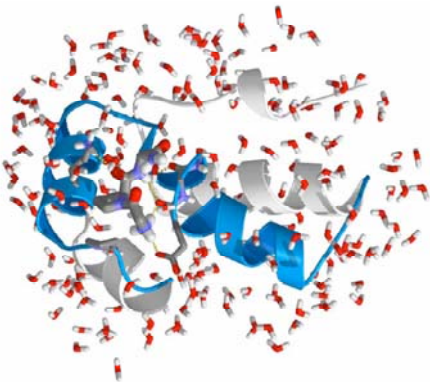
# 参考书目



M. P. Allen and D. J. Tildesley, Computer Simulation of Liquids  
(Clarendon Press, Oxford, 1987)

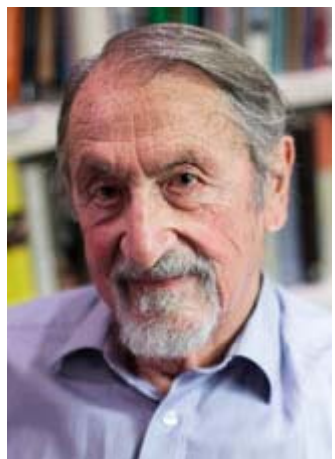


D. Frenkel and B. Smit, Understanding Molecular Simulation  
(Academic Press, San Diego, 2002)



Gromacs User Manual

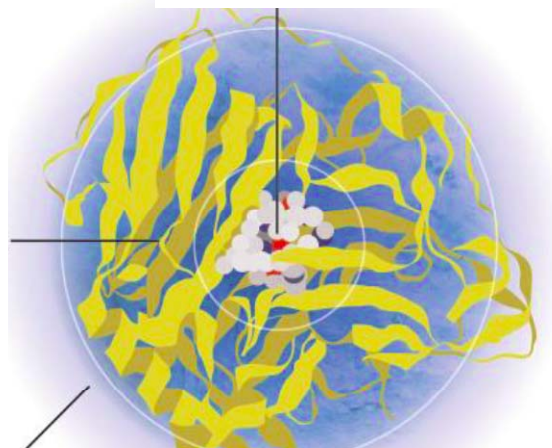
# 2013诺贝尔化学奖：分子动力学模拟，多尺度模型



Martin Karplus Michael Levitt Arieh Warshel

量子力学

经典力学

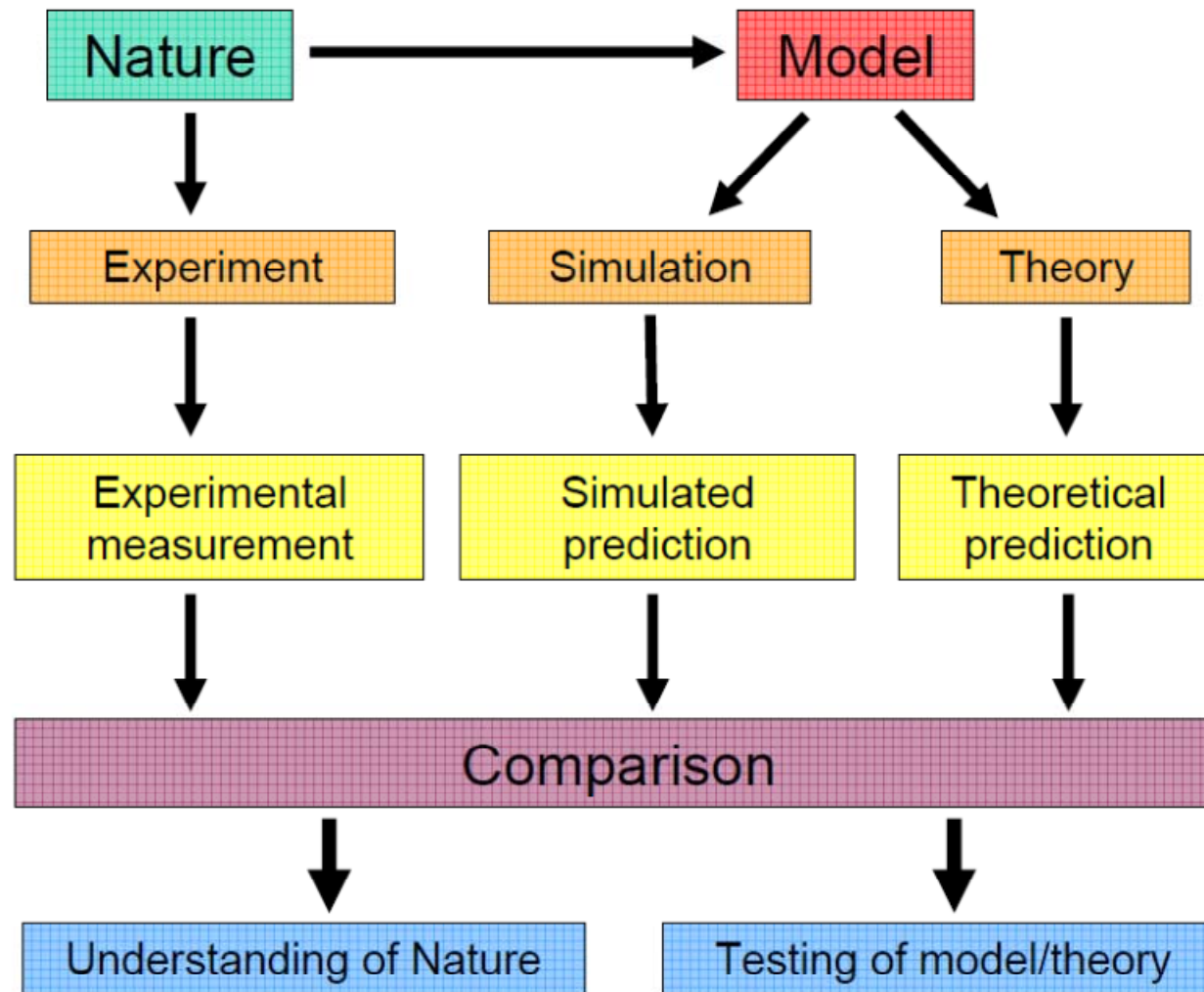


连续介质模型

“发展了用于研究复杂化学体系的多尺度模型”

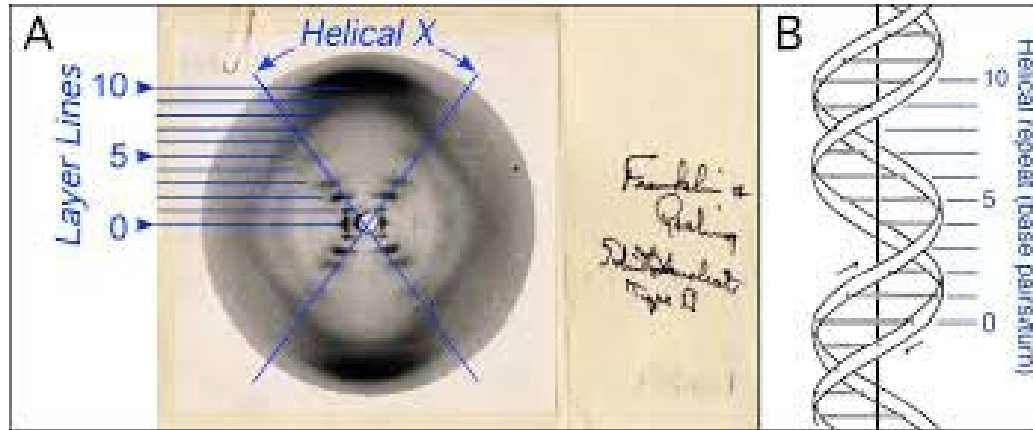
“这些在计算机上进行的实验极大地加深了人们对化学过程的认识”

# 理论模型的作用

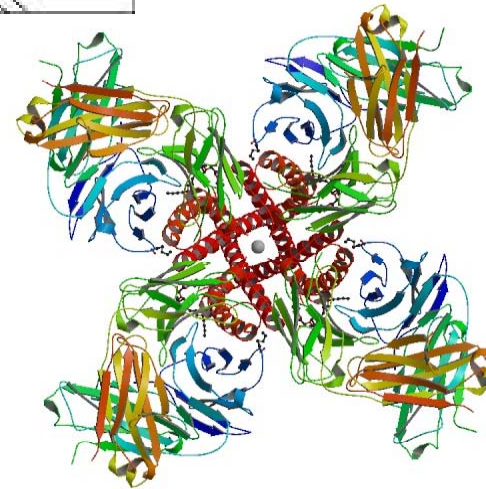
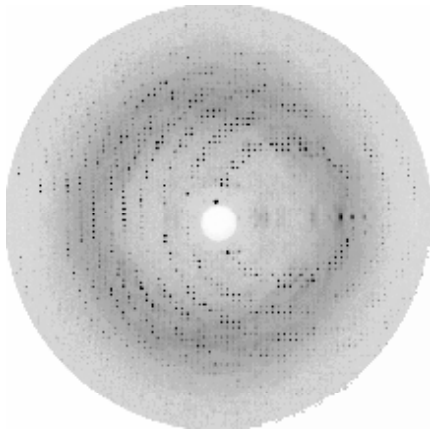


好的理论模型对于**认识**和**预测**实验现象作用巨大

# 例子：X射线衍射



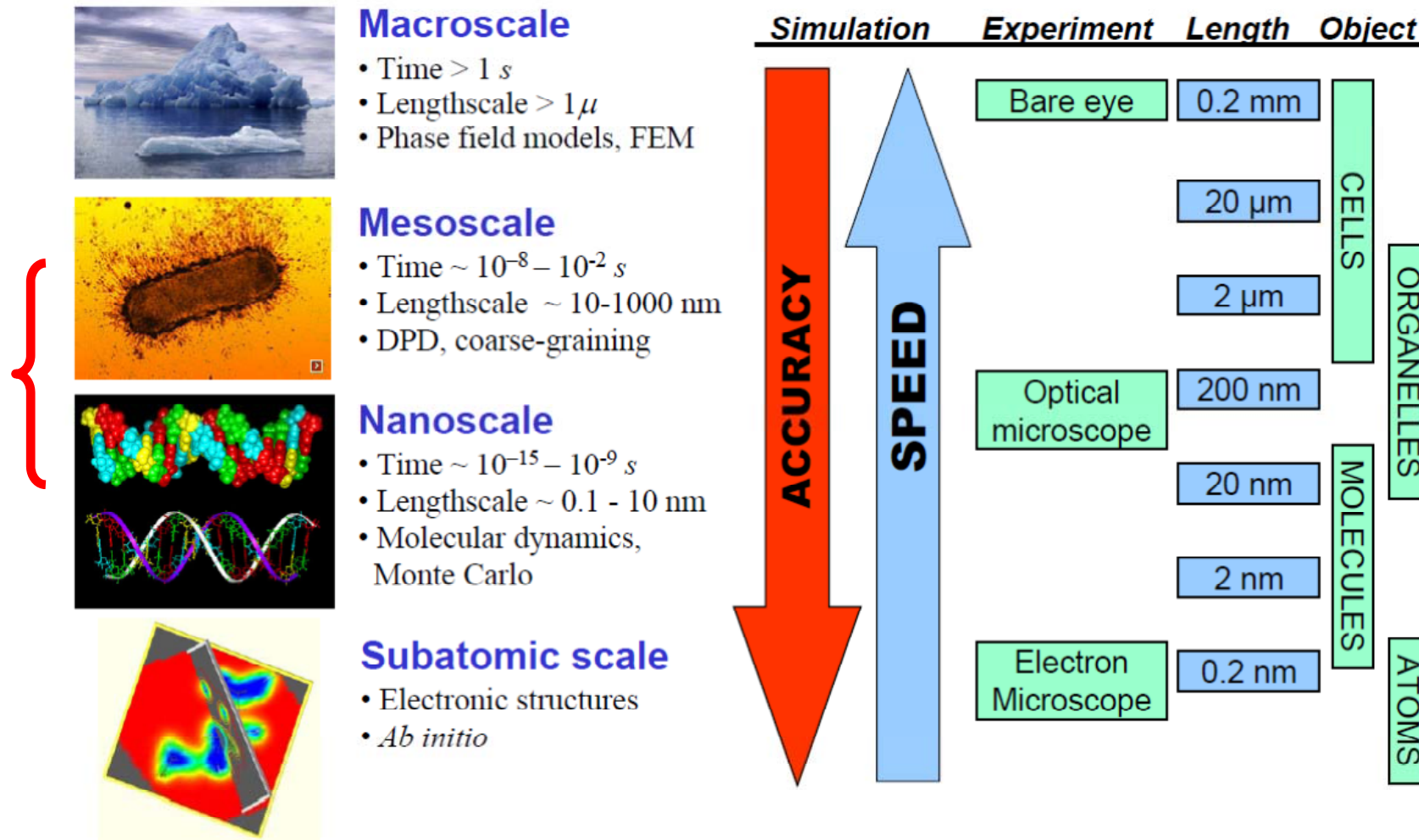
**DNA Structure,  
1962 Nobel Prize in  
Physiology or  
Medicine**



**R. MacKinnon, 2002 Nobel Prize in Chemistry**

**背后的理论化学模型：力场模型、结构优化**

# 时间与空间尺度



所有模拟都是精度和计算可行性的折中

# 经典还是量子动力学

- |                                               |   |                                                                              |
|-----------------------------------------------|---|------------------------------------------------------------------------------|
| 1. polypeptide folding                        | } | <i>thermodynamic equilibria<br/>governed by weak<br/>(non-bonded) forces</i> |
| 2. biomolecular association                   |   |                                                                              |
| 3. partitioning between solvents              |   |                                                                              |
| 4. membrane/micelle formation                 |   |                                                                              |
| 5. chemical reactions, enzyme catalysis       | } | <i>chemical transformations<br/>governed by strong forces</i>                |
| 6. enzyme catalysis                           |   |                                                                              |
| 7. photochemical reactions, electron transfer |   |                                                                              |

## *characteristics (1-4):*

- |                        |                           |   |              |
|------------------------|---------------------------|---|--------------|
| - degrees of freedom:  | atomic (solute + solvent) | → | classical MD |
| - equations of motion: | classical dynamics        |   |              |
| - governing theory:    | statistical mechanics     |   |              |

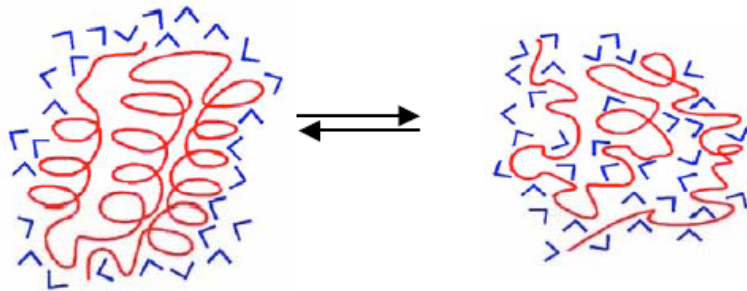
## *characteristics (5-7):*

- |                        |                               |   |            |
|------------------------|-------------------------------|---|------------|
| - degrees of freedom:  | electronic, nuclear           | → | quantum MD |
| - equations of motion: | quantum dynamics              |   |            |
| - governing theory:    | quantum statistical mechanics |   |            |

### Folding

folded/native

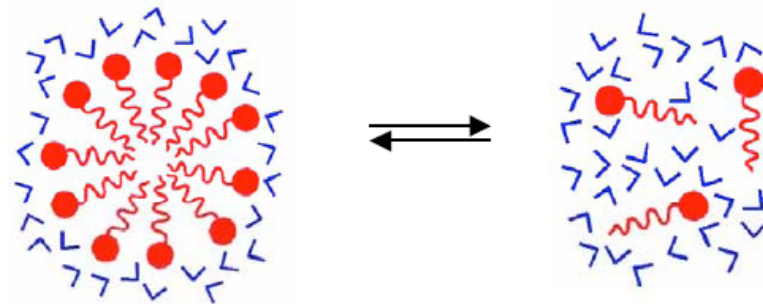
denatured



### Micelle Formation

micelle

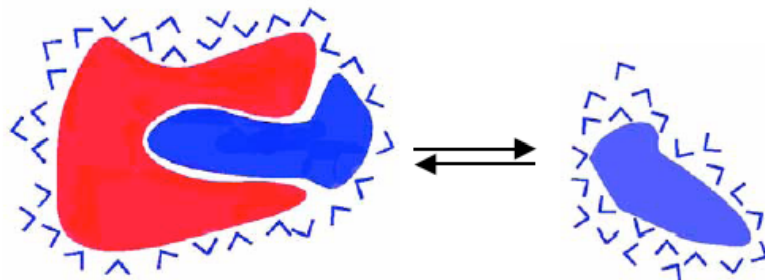
mixture



### Complexation

bound

unbound

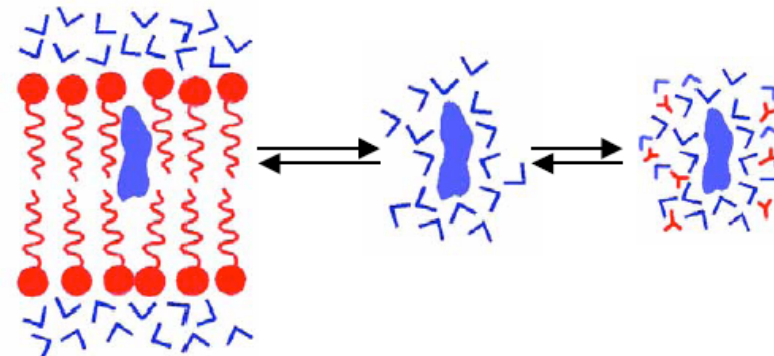


### Partitioning

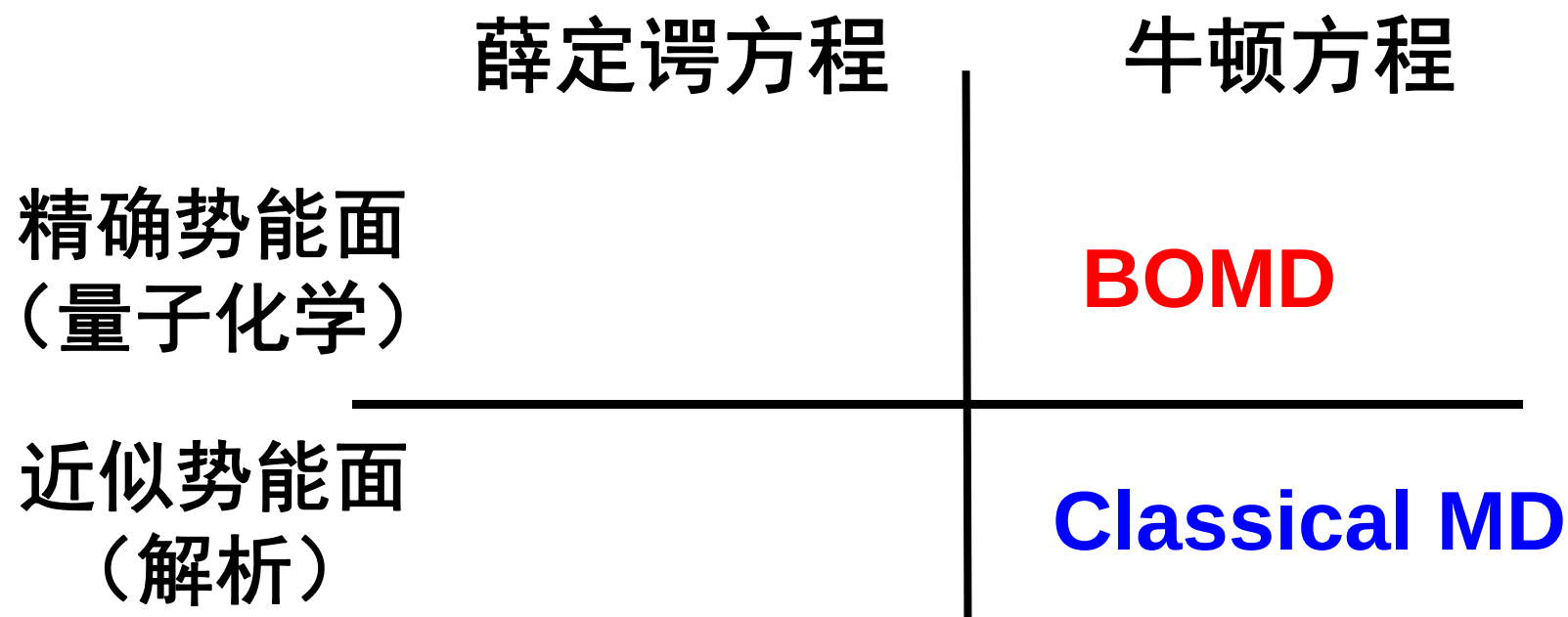
in membrane

in water

in mixtures



# 经典分子动力学模拟

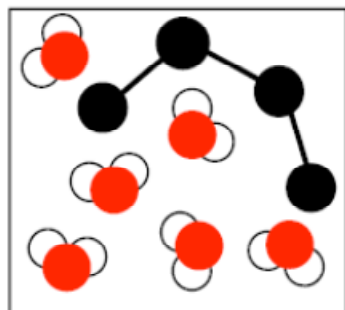


其它: QM/MM, non-adiabatic MD,...

# 经典分子动力学模拟的基本要素

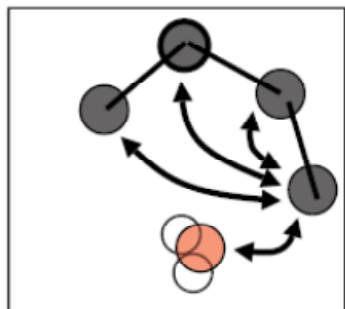
Every molecule consists of atoms that are very strongly bound to each other

如何描述分子



*Degrees of freedom:  
atoms as elementary  
particles + topology*

*Forces or  
interactions  
between atoms*

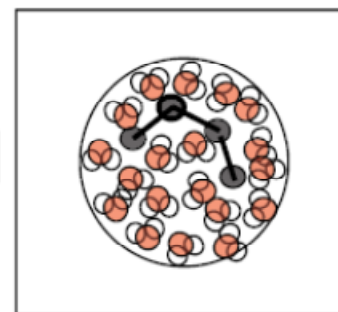


Force field =  
physico-chemical  
knowledge

相互作用

如何模拟外界条件

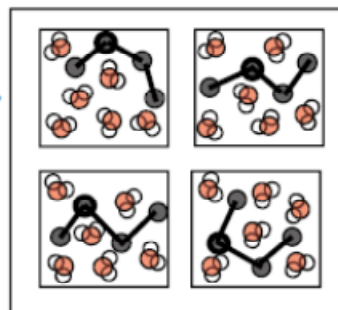
*Boundary conditions*



system  
temperature  
pressure  
walls  
external forces

*Methods to generate  
configurations of  
atoms: Newton*

随时间演化



**MOLECULAR  
MODEL**

# 分子模型：质点、弹簧...

